Chemical modification studies of the active centre of *Candida albicans* chitinase and its inhibition by allosamidin

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Allosamidin, a glycoside antibiotic, is shown to be a strong, competitive inhibitor of semi-purified chitinase from yeast cells of *Candida albicans*. The inhibitory potency of allosamidin was pH-dependent, with IC50 values of 280 nM at pH 5.0 and 21 nM at pH 7.5. At higher, micromolar, concentrations, allosamidin inactivated this chitinase in a time- and concentration-dependent manner. Kinetic studies of this inactivation provided evidence for the formation of a reversible complex between allosamidin and chitinase, characterized by *Km* values of 5 μM, followed by irreversible modification of the enzyme with velocity constant *k*₂ = 4.6 × 10⁻³ s⁻¹. Chemical modification studies with the use of group-specific reagents suggested the presence of Glu/Asp carboxyl group(s) at or near the active site, that were important for enzyme activity. The carboxyl-specific reagent, 1-ethyl-3-(3-dimethylaminopropyl)-carbodiimide, inactivated the chitinase in a single step process, with apparent second-order rate constant of 0.014 M⁻¹ s⁻¹.

Introduction

Chitinase is an enzyme of widespread occurrence with a wide range of roles (Gooday, 1991). In the yeast *Saccharomyces cerevisiae*, it is involved in separation of cells after budding (Kuranda & Robbins, 1991; Cabib et al., 1992) and in filamentous fungi it is probably involved in hyphal growth and branching and in spore germination (Gooday et al., 1986; Rast et al., 1991). Chitinase activity has been described from yeast cells of the dimorphic human pathogenic fungus *Candida albicans* (Barrett-Bee & Hamilton, 1984; Dickinson et al., 1989). Because of its roles in fungal growth, chitinase presents a target for potential anti-fungal agents. A specific inhibitor of some chitinases has recently been characterized, as a novel antibiotic from *Streptomyces* species, and has been named allosamidin (Sakuda et al., 1987) or A82516 (Somers et al., 1987). The chitinase from *C. albicans* is susceptible to allosamidin (Dickinson et al., 1989). We present here results of a study of the mode of action of the chitinase of *C. albicans*, as elucidated by the effects of selected group specific reagents and of allosamidin on a semi-purified preparation.

* Abbreviations: EDC, 1-ethyl-3-(3-dimethylaminopropyl)-carbodiimide; (GlcNAc)₃, N,N'-diacetylchitobiose; (GlcNAc)₄, N,N',N"-triacetylchitotriose; (GlcNAc)₅, N,N',N"',N""-tetracetylchitotetraose; MeUmb, 4-methylumbelliferyl; p-CIMB, parachloromercuribenzoate.

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solution. After 30 min incubation at 37°C, 0.05 ml 5 M-NaOH was added to each well to stop the reaction. Fluorescence was measured immediately in the spectrophotometer at excitation band 363 nm and absorption band 440 nm. The concentration of MeUmb released was read from the standard curve. Protein was assayed by the method of Bradford (1976) with BSA as a standard.

**Determination of kinetic parameters of inhibition.** Initial velocities of chitinase-catalysed hydrolysis of MeUmb(GlcNAc), were determined by continuous monitoring of MeUmb release, in samples containing various concentrations of the substrate (1.25–20 μM) and allosamidin (0–300 nM). Lineweaver–Burk plots were constructed using least square method.

**Inactivation of chitinase by allosamidin and chemical modification of the enzyme.** Chemical modifications of chitinase were performed using a variety of group-selective reagents and allosamidin. Buffers for incubation mixtures were chosen to ensure optimal conditions of inactivation, according to literature data: chemical modification by thiol-binding reagents and by PMSF were performed in 0.1 M-PIPES, pH 7.5; participation of histidine residues was investigated by treatment with diethylpyrocarbonate in 0.1 M-MES, pH 6.0; phenylglyoxal was used as an agent specifically reacting with the guanidyl group of arginine, in 0.1 M-PIPES, pH 6.0; modification of the carboxyl group of Asp/Glu residues was performed using EDC in 0.1 M-PIPES, pH 6.0; inactivation by allosamidin was studied in different buffer systems but optimal conditions were achieved in 0.1 M-PIPES, pH 7.5.

In all inactivation experiments, 1 μg of enzyme protein containing the semi-purified chitinase was incubated at 25°C in buffer, with a given inactivator at appropriate concentration (total volume 1 ml). The gel-filtration-centrifuge method (Penefsky, 1977) was used to follow the inactivation, as follows: 0.2 ml samples of incubation mixtures were periodically withdrawn and applied at tops of 1 ml columns packed with Sephadex G-50 (previously equilibrated with an appropriate buffer) and centrifuged (1 min, 200 g). Under these conditions the unbound inhibitor was separated from the enzyme and the protein was recovered in clean test tubes. Appropriate effluent samples were used for the determination of the residual enzyme activity by using the standard assay method. The efficiency of this technique for the separation of protein from unbound inhibitor was tested by three methods: (i) Three samples of 0.2 ml of a solution containing Ellman’s reagent (100 μM) and BSA (100 μg ml⁻¹) were treated as described. Under these conditions, Ellman’s reagent was retained in the gel bed as a clearly visible narrow yellow band just below the top of the gel. The protein concentrations of the eluates were 100 ± 3 μg ml⁻¹. (ii) Samples of 0.2 ml of a solution of allosamidin (10 μM) and BSA (100 μg ml⁻¹) were treated as described. The possible presence of allosamidin in the eluates was tested by assaying for inhibition of chitinase (pH 7.5, 0.1 M-PIPES buffer). No inhibition was observed, while in a simultaneous experiment, 1 nM-allosamidin in the presence of BSA (100 μg ml⁻¹) gave 40 ± 3% inhibition of chitinase activity. BSA alone did not affect chitinase activity. (iii) Protein present in the chitinase preparation was recovered with 100% efficiency after treatment as described.

**Results**

MeUmb(GlcNAc), was chosen as a soluble substrate for determination of C. albicans chitinase activity. The enzyme exhibited much higher activity towards this substrate than towards other MeUmb(GlcNAc), oligomers (dimer, trimer). This suggests that its major activity is as an endochitinase (cf. McCreath & Gooday, 1992). It had no effect on MeUmb-GlcNAc, so had no N-acetylglucosaminidase activity. Using MeUmb(GlcNAc), we confirmed the strong inhibitory potency of allosamidin against this enzyme. The effectiveness of this inhibition was dependent on pH (Fig. 1). The IC₅₀ value at pH 7.5 was several-fold lower than that at pH 5.0 (21 μM and 280 μM, respectively). The mode of inhibition was competitive with inhibition constants as low as 10 nm in 0.1 M-PIPES, pH 7.5 and 62.5 nm in 0.1 M-PIPES, pH 6.0 (Fig. 2). Similar values and inhibition were obtained in MES and phosphate buffers. Surprisingly, results obtained in McIlvaine’s citrate/phosphate buffer suggested mixed or even non-competitive mode of inhibition (data not shown).

Incubation of partially purified chitinase with allosamidin in the absence of substrate led to the irreversible inactivation of the enzyme. Allosamidin (10 μM) inactivated chitinase completely after 20 min incubation. The time course of inactivation is shown in Fig 3. The irreversibility of this phenomenon was shown by incubating chitinase (1 μg ml⁻¹) with allosamidin (10 μM) for 30 min at 37°C, isolating the allosamidin-enzyme complex by the gel-filtration-centrifuge method, reincubating this for a further 30 min at 37°C, and then subjecting it to a further cycle of the gel-filtration-centrifugation procedure. There was no recovery of chitinase activity. When values of apparent rate constants of inactivation (kapp), calculated from the slope of lines in Fig. 3, were plotted against allosamidin
concentration, a hyperbolic curve was obtained (not shown). Such a pattern is consistent with the formation of a reversible complex before inactivation:

$$E + I \rightleftharpoons [E*I] \rightarrow EI^*$$

[E*I] is the enzyme-inhibitor complex and EI* is the irreversibly modified enzyme.

Assuming that [I] $\gg$ [E] and that the reversible complex is at all times in equilibrium with enzyme and inhibitor, the equation derived by Meloche (1967) can be applied:

$$\tau = \frac{1}{[I]} (T \cdot K_{\text{inact}} + T)$$

where $K_{\text{inact}}$ is the inactivation constant, $\tau$ is a particulate half-time of inactivation and $T$ is the minimal inactivation half-time at infinite inhibitor concentration.

Values of $K_{\text{inact}}$, $T$ and $k_2$ were calculated from a linear plot of inactivation half-times versus reciprocal of allosamidin concentration (Fig. 4). The values obtained were:

$$K_{\text{inact}} = 5 \mu M; \; T = 2.5 \text{ min}; \; k_2 = \frac{\ln 2}{\tau} = 4.6 \times 10^{-3} \text{ s}^{-1}$$

The calculated value of apparent second-order rate constant ($k_2/K_{\text{inact}}$) was $0.92 \times 10^3 \text{ M}^{-1} \text{ s}^{-1}$, i.e. much lower than $k_1$ (commonly at least $10^7 \text{ M}^{-1} \text{ s}^{-1}$ for substrates). Therefore the quasi-equilibrium assumption for the kinetic two-step model of inactivation is fully justified (Brocklehurst, 1979).

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Fig. 2. Inhibition of chitinase by allosamidin. Incubation mixtures consisted of the enzyme protein (1 $\mu$g ml$^{-1}$), MeUmb(GlcNAc)$_4$ (0–20 $\mu$M) and allosamidin ($\Delta$, 0 $\mu$M; $\bigcirc$, 25 $\mu$M; $\blacklozenge$, 50 $\mu$M; $\nabla$, 100 $\mu$M) in 0.2 ml portions of 0.1 M-PIPES buffer, pH 6.0. The reaction was carried out, stopped and the fluorescence measured as described in the legend to Fig. 1. The results were plotted as $1/V$ versus $1/[S]$. All points are the average of three determinations.

Fig. 3. Time course of inactivation of chitinase by allosamidin. Enzyme protein (1 $\mu$g ml$^{-1}$) was incubated with allosamidin ($\Delta$, 0.25 $\mu$M; $\bigcirc$, 0.5 $\mu$M; $\blacklozenge$, 1.0 $\mu$M; $\Delta$, 2.0 $\mu$M; $\bigcirc$, 10 $\mu$M) in PIPES buffer, pH 7.5 (total vol. 1 ml). Samples (0.2 ml) were periodically withdrawn, applied to the tops of columns packed with Sephadex G-50, and centrifuged. The residual enzyme activity was assayed in effluent samples by the fluorimetric method.

Fig. 4. Protective effect of (GlcNAc)$_n$ oligomers against allosamidin-mediated inactivation of chitinases. Mixtures containing enzyme protein (1 $\mu$g ml$^{-1}$), allosamidin and (GlcNAc)$_n$ oligomers (1 mM) were incubated at 25°C. Samples (0.2 ml) were periodically withdrawn and treated as described in the legend to Fig. 3. Data obtained from the time course of inactivation (not shown) were plotted as $\tau = f(1/[I])$. $\nabla$, allosamidin alone; $\bigcirc$, allosamidin + (GlcNAc)$_4$; $\blacklozenge$, allosamidin + (GlcNAc)$_2$. The same graph shows the protection of chitinase against inactivation by allosamidin provided by short (GlcNAc)$_n$ oligomers. The protective effect of (GlcNAc)$_4$ was much stronger than those of (GlcNAc)$_2$ and (GlcNAc)$_3$. GlcNAc had no effect (not shown).
Table 1. Chemical modification of C. albicans chitinase by group-specific reagents

<table>
<thead>
<tr>
<th>Group</th>
<th>Reagent</th>
<th>Inactivation</th>
</tr>
</thead>
<tbody>
<tr>
<td>1. -SH (Cys)</td>
<td>Iodoscetic acid (10 mM)</td>
<td>46% after 30 min</td>
</tr>
<tr>
<td></td>
<td>Iodoacetamide (10 mM)</td>
<td>52% after 30 min</td>
</tr>
<tr>
<td></td>
<td>p-CIMB (5 mM)</td>
<td>58% after 30 min</td>
</tr>
<tr>
<td></td>
<td>N-Ethylmaleimide (10 mM)</td>
<td>30% after 30 min</td>
</tr>
<tr>
<td>2. -OH (Ser/Thr)</td>
<td>PMSF (1 mM)</td>
<td>None</td>
</tr>
<tr>
<td>3. -guanidyl (Arg)</td>
<td>Phenylglyoxal (50 mM)</td>
<td>None</td>
</tr>
<tr>
<td>4. -imidazoyl (His)</td>
<td>Diethylpyrocarbonate</td>
<td>None</td>
</tr>
<tr>
<td>5. -carboxyl (Glu/Asp)</td>
<td>EDC</td>
<td>Dependent on time and concentration</td>
</tr>
</tbody>
</table>

Several group-specific chemical reagents were used to investigate amino acids at the active site of chitinase. The results of these studies are summarized in Table 1. Most of the agents used failed to inactivate the enzyme. Partial inactivation was observed with thiol-blocking reagents, but this effect was seen only at very high concentrations. On the other hand, time- and concentration-dependent inactivation by the carboxyl-specific carbodiimide derivative EDC was demonstrated. Semilogarithmic plots of residual activity as a function of time were biphasic, indicating complexity of the EDC-induced inactivation (Fig. 5). However, after resolving into two first-order processes, the slope at longer times, obviously determining the rate of inactivation, could be used for calculation of apparent pseudo-first-order constants. A plot of these constants against EDC concentration was linear (Fig. 6) and provided a second-order rate constant of $14 \times 10^{-3} \text{ M}^{-1} \text{s}^{-1}$. Analysis of the order of inactivation with respect to EDC concentration by the method of Levy et al. (1963) yielded a slope of 0.91 (not shown), indicating that one molecule of EDC binds to one molecule of enzyme when inactivation occurs.

(GlcNAC)$_n$ oligomers protected chitinase against inactivation by EDC (Fig. 6). However, in this case, (GlcNAC)$_3$ was nearly as good as (GlcNAC)$_4$ as a protective agent; both were much better than (GlcNAC)$_6$.

Discussion

Allosamidin, a pseudo-trisaccharide containing N-acetyllallosamine (Sakuda et al., 1987; Somers et al., 1987), is a strong inhibitor of chitinases from the insect
**The stoichiometry of chitinase**

The presence of serine, histidine and arginine at the active modification studies could provide some evidence for the possible participation of particular amino acids in nucleophilic residue at the enzyme active site. Currently, was not determined; it should be 1 if the inhibitor is to return upon gel filtration; inactivation obeyed saturation kinetics and substrates [in this case (GlcNAc), oligomers] protected the enzyme against inactivation. The stoichiometry of chitinase:allosamidin interaction was not determined; it should be 1:1 if the inhibitor is to be considered an active-site-directed agent. Nevertheless, the structural analogy between allosamidin and (GlcNAc)ₙ oligomers – good substrates for chitinase – makes this assumption very likely.

Since the allosamidin-chitinase adduct was not isolated and its structure not determined, one can only speculate about the possible mechanism of inactivation. The aminocylitol part of allosamidin may well behave as an isourea derivative and therefore act as an alkylating agent. If so, a covalent bond could be formed with any nucleophilic residue at the enzyme active site. Currently, little is known of the active site of chitinase. Chemical modification studies could provide some evidence for the possible participation of particular amino acids in catalytic action. Data presented in this work exclude the presence of serine, histidine and arginine at the active site of chitinase. On the other hand, it seems very likely that Glu/Asp carboxyl group(s) play an important role in the catalytic centre of the enzyme. Such a situation is common among other glycosidases like lysozyme (Lin & Kosshland, 1969), β-glucosidase (Clarke, 1990) and cellulase (Moranelli et al., 1986). By hydrophobic cluster analysis of amino acid sequences from a range of plant and bacterial chitinases, Henrissat (1990) reports the presence of four invariant acid residues, three aspartate and one glutamate, in the putative active-site region. Further analysis of more sequences (B. Henrissat, personal communication) has reduced the number of putative catalytic residues to two, an aspartate and a glutamate. Recently Kuranda & Robbins (1991) cloned and sequenced chitinase from *S. cerevisiae*. By comparison with other chitinase sequences, these authors also suggested the presence of the three aspartate and one glutamate residues in the catalytic domain of this enzyme. Since *C. albicans* and *S. cerevisiae* are related micro-organisms, one should expect a high extent of homology in sequences of respective enzymes.

The *S. cerevisiae* chitinase also has a chitin-binding domain, rich in cysteine residues (Kuranda & Robbins, 1991). We observed the partial inactivation of *C. albicans* chitinase by thiol-blocking reagents. Since the substrate used by us for the determination of enzyme residual activity is shorter than natural chitin chains, it is possible that blocking of the cysteine residues at the binding domain does not prevent MeUmb(GlcNAc)ₙ from gaining access to the catalytic site. In any case, high concentrations of thiol reagents necessary for the partial inactivation of chitinase exclude the participation of any cysteine residue in the catalytic action of the enzyme. It should be added that some other chitinases have been reported to be inhibited by Hg²⁺ ions (Somers et al., 1987) and iodoacetic acid (Yabuki et al., 1986). Both these agents are known as thiol-blocking compounds. However, divalent metal ions are also able to form stable complexes with carboxylic groups at the active sites of enzymes (Perkins et al., 1979). On the other hand, halogeno derivatives of aliphatic carboxylic acids have been reported to alkylate nucleophilic carboxylate anions, as was shown in bromopyruvate mediated inactivation of aldolase (Vlahos & Dekker, 1990).

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**References**


