Characterization and partial purification of L-asparaginase from Corynebacterium glutamicum

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A high L-asparaginase (L-asparagine amidohydrolase; EC 3.5.1.1) activity was found under conditions of lysine overproduction in cultures of Corynebacterium glutamicum. L-Asparaginase was purified 98-fold by protamine sulphate precipitation, DEAE-Sephasel anion exchange, ammonium sulphate precipitation and Sephacryl S-200 gel filtration. The asparaginase protein was subjected to PAGE under non-denaturing conditions, identified by an in situ reaction and eluted from the gel in an active form. The estimated Mf, from gel filtration and SDS-PAGE was 80000. The L-asparaginase activity was inhibited by the L-asparagine analogue 5-diazo-4-oxo-L-norvaline. Neither D-asparagine nor L-glutamine was a substrate for the enzyme. L-Asparaginase was produced constitutively; its role may be that of an overflow enzyme, converting excess asparagine into aspartic acid, the direct precursor of lysine and threonine.

Introduction

L-Asparaginase (L-asparagine amidohydrolase; EC 3.5.1.1) catalyses the hydrolysis of L-asparagine to L-aspartic acid and ammonia. In the presence of hydroxylamine this enzyme converts the activated L-asparagine into aspartyl β-hydroxamate (Wriston, 1970). This activity of the asparaginase is routinely used for the determination of enzyme activity.

Asparaginase has been studied in Escherichia coli and other Gram-negative bacteria such as the Achromobacteriaceae (Roberts et al., 1972) and Vibrio succinogenes (Kafkewitz & Goodman, 1974).

Amino-acid-producing corynebacteria are of great industrial interest because they excrete large amounts of various amino acids (Martin et al., 1987; Martin, 1989). Lysine, threonine and methionine, three of the commercially important amino acids produced by Corynebacterium glutamicum, are derived from aspartic acid, which, under normal physiological conditions, might be limiting for lysine and/or threonine biosynthesis. Aspartic acid is formed from either Krebs’ cycle intermediates using glutamic acid as the amino donor or from asparagine by the action of asparaginase. A very active L-asparaginase was found in C. glutamicum (in preliminary experiments) under lysine-producing fermentation conditions. In addition to their role in amino acid metabolism, several asparaginases are of interest because of their antitumour properties (Handschumacher et al., 1968; Kafkewitz & Goodman, 1974; Roberts et al., 1972; Willis & Woolfolk, 1974). The asparaginases of Gram-positive bacteria have received little attention. Therefore, it was of interest to purify the L-asparaginase of C. glutamicum and to study its kinetics as a first step towards understanding its role in the biosynthesis of the aspartic acid family of amino acids.

Methods

Micro-organisms and growth conditions. Corynebacterium glutamicum ATCC 13032 (Abe et al., 1967), initially named Micrococcus glutamicus (Kinoshita et al., 1958), was used in this study. Cultures were grown aerobically in tryptone-soya broth (TSB) medium containing (g l⁻¹): casein peptone, 17; soya peptone, 3; glucose, 2.5; NaCl, 5; K₂HPO₄, 2.5; yeast extract, 1, pH 7.3. Cultures were incubated at 30°C (50 ml in 250 ml flasks) in a New Brunswick Scientific orbital incubator at 280 r.p.m. for 24 h.

Cell-free extracts. Cells were grown for 24 h in TSB medium (to the end of the exponential phase), collected by centrifugation at 10000 g for 15 min and washed twice with 0.9% NaCl at 4°C. Washed cells (about 8 g wet wt per litre of culture) were either used immediately for enzyme purification or frozen at -80°C for later use. Cells were disrupted by sonification (Kase & Nakayama, 1974) as follows. Cells (5 g wet wt) were suspended in 20 ml of TM buffer (0.1 M-Tris/HCl, pH 8.0; 0.02 M-β-mercaptoethanol). Sonication was done for periods of 30 s with 1 min intervals in an ice-cooled tube using a Branson sonifier (model B-12) at 75–100 W, until the cells were completely disrupted as observed microscopically. Cell debris was removed by centrifugation at 35000 g for 30 min. The clear supernatant (S15) was dialysed overnight against...
TM buffer or filtered through Sephadex G-25 equilibrated with TM buffer. Enzyme extraction and purification were done at 0-4°C.

The dialysed S35 extract was used directly for enzyme assays or purified further (see below). The activity of S35 extracts frozen at -80°C remained unchanged for at least 8 weeks. However, repeated thawing and freezing led to an irreversible loss of activity.

Purification of L-asparaginase. L-Asparaginase was partially purified in four sequential steps. Dialysed S35 extracts were mixed slowly with 0·1 vol. 1% (w/v) protamine sulphate (Sigma) in TM buffer with gentle agitation for 20 min. The nucleic acid precipitate was removed by centrifugation at 15000 g for 20 min and the supernatant was chromatographed through a DEAE-Sephadex (Pharmacia) anion exchange column equilibrated with TM buffer. The column was washed with 200 ml of TM buffer and the proteins eluted with 200 ml of a linear gradient (0-0.5 M) of ammonium sulphate in TM buffer. The active fractions were collected, pooled and fractionated with increasing concentrations of ammonium sulphate (Sigma; enzyme grade). Most of the L-asparaginase activity precipitated in the 30-45% saturation fraction. The enzyme precipitate was dissolved in TM buffer and filtered through a Sephacryl S-200 or Sephadex G-200 column (2·6 × 100 cm), eluting with TM buffer at a flow rate of 0·5 ml min⁻¹.

The M₉ of L-asparaginase was calculated by comparison with globular proteins of known M₉ (Pharmacia calibration kit). Partially purified preparations kept their activity when frozen at -80°C but lost 50% of their activity in 3 d at 4°C.

Protein concentration in the extracts was determined by the Lowry method, using BSA as standard.

Assay of L-asparaginase activity using hydroxylation. The reaction mixture contained 25 µmol L-asparagine, 400 µmol hydroxylamine hydrochloride (neutralized to pH 7.0 with NaOH), 0·05-0·02 units of enzyme and 100 µmol Tris/HCl, pH 7.0, in a final volume of 1·0 ml. The reaction mixture was incubated at 40°C for 20 min except in temperature effect studies. The reaction was stopped by addition of 1·5 ml of chromogenic iron chloride reagent [FeCl₃ (100 g) and trichloroacetic acid (33 g) in 1 litre 0·7 M-HCl] and centrifuged at 1800 g. The aspartyl β-hydroxamate formed was quantified spectrophotometrically at 515 nm by comparison with a standard curve of pure aspartyl β-hydroxamate (Sigma). Control reactions without substrate (L-asparaginase) or with heat-inactivated enzyme (10 min at 100°C) were run in each experiment. An enzyme unit (U) was defined as the enzyme activity that forms 1 µmol aspartyl β-hydroxamate min⁻¹ at 40°C. Specific activities are given as mU (mg protein)⁻¹.

HPLC assay of L-asparaginase activity. The reaction mixture, temperature and time were identical to those used in the hydroxylamine assay but hydroxylamine was not added to the reaction mixture. The reaction product, L-aspartic acid, was quantified by HPLC. The reaction was stopped by addition of 1 vol. of methanol and the reaction mixture was cooled in ice and centrifuged at 1800 g. The amino acids in the supernatant were derivatized with p-hthalaldehyde (OPA) according to the procedure of Jones et al. (1981). OPA-labelled amino acids were quantified after HPLC separation in a Varian 5000 chromatograph equipped with a Waters 420 AC Fluorescence detector, following the method of Vuillet et al. (1986) modified as follows: a reverse phase C18 µ-Bondapak column (Waters) was used and the amino acids were eluted with 20% (v/v) methanol in 0·02 M-potassium phosphate buffer, pH 7·0, at a flow rate of 1·5 ml min⁻¹.

PAGE. Non-denaturing PAGE of purified (98-fold) L-asparaginase was done essentially as described by Laemmli (1970) except that SDS was omitted. Electrophoresis was done at 4°C in a vertical slab gel (170 × 130 × 1·5 mm; 10% w/v, polyacrylamide and 0·26%, N,N'-methylene-bis-acrylamide as cross-linking agent (Hjerten, 1962)) at 100 V and 60 mA.

After electrophoresis, part of the gel was stained with Coomassie brilliant blue R-250 and the other was used for in situ detection of L-asparaginase. This part of the gel was submerged in reaction mixture containing hydroxylamine and incubated at 40°C for 30 min. The reaction was stopped and the reaction product stained by addition of 1:5 vols of the iron chloride reagent. An ochre band was clearly visible in the gel.

The L-asparaginase was eluted in an active form from unstained preparative non-denaturing gels developed as above, and was used for further studies. Some samples of enzyme were used for M₉ determination by SDS-PAGE, done as described by Laemmli (1970).

Chemicals. L- and D-amino acid substrates were obtained from Sigma or Merck. The protein calibration kit, Sephacryl S-200 and Sephadex G-200 were from Pharmacia. 5-Diazo-4-oxo-L-norvaline (DONV), an asparaginase analogue, was kindly provided by R. E. Handschumacher, Yale University, USA. All other chemicals were of reagent quality.

Results

L-asparaginase activity in cell-free extracts

Dialysed crude extracts showed high L-asparaginase activity using the hydroxylamine assay. Similar levels of L-asparaginase activity [about 20-21 mU (mg protein)⁻¹] were found after growth of C. glutamicum in TSB medium and in minimal medium. The enzyme was specific for L-asparagine and could not use D-asparagine or the D- or L-isomers of aspartic acid, glutamic acid and glutamine as substrates. However, non-dialysed extracts showed activity with L-aspartic acid as substrate due to the presence in the extracts of aspartokinase and aspartase activities.

The final product of the reaction mixtures without hydroxylamine was characterized by HPLC. Under these reaction conditions a clear peak of L-aspartic acid (retention time 2·5 min) was observed. The reaction proceeded with quantitative conversion of L-asparagine (retention time 9·0 min) into L-aspartic acid. About 90-95% of the L-asparagine was converted to L-aspartic acid after 30 min incubation. Similar apparent enzyme activities were obtained using HPLC analysis and the hydroxylamine method.

Optimal L-asparaginase activity was found at pH 7·0 and 40°C. The enzyme activity was slightly lower at pH values of 7·5 or 8·0 and decreased markedly at pH 6·5. There was almost no activity at pH 6·0. The reaction rate was linear for 30 min under optimal pH and temperature conditions, and also with increasing enzyme concentrations up to 0·2 U of enzyme per ml of reaction mixture. The optimal parameters of the L-asparaginase did not change when the enzyme was purified.

Purification of L-asparaginase

L-Asparaginase was purified from 1554 mg of soluble protein in 37 ml of dialysed S35 extract. After removal of
Table 1. Partial purification of L-asparaginase from C. glutamicum

<table>
<thead>
<tr>
<th>Purification step</th>
<th>Vol. (ml)</th>
<th>Total protein (mg)</th>
<th>Total activity (U)</th>
<th>Specific activity [mU (mg protein)^-1]</th>
<th>Recovery (%)</th>
<th>Purification (-fold)</th>
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<tr>
<td>Cell extract</td>
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<td>1554</td>
<td>32</td>
<td>20.6</td>
<td>100</td>
<td>1</td>
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<tr>
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<td>33.6</td>
<td>24.7</td>
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<tr>
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<td>14</td>
<td>82.4</td>
<td>11.2</td>
<td>136</td>
<td>35</td>
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</tr>
<tr>
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<td>5.1</td>
<td>335.8</td>
<td>16</td>
<td>21.3</td>
</tr>
<tr>
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<td>5-5</td>
<td>2</td>
<td>4</td>
<td>2020</td>
<td>12.5</td>
<td>98</td>
</tr>
</tbody>
</table>

nucleic acids by protamine sulphate precipitation, the supernatant was passed through an anion exchange DEAE-Sepharose column. Elution with a linear gradient of ammonium sulphate gave a single symmetrical peak of L-asparaginase activity at 270 mM-ammonium sulphate. The purification obtained in this step was 6-6-fold (Table 1). The active fractions were pooled, dialysed against TM buffer and fractionated with ammonium sulphate. L-Asparaginase was collected in the 30-45% ammonium sulphate fraction with a specific activity of 335-6 mU (mg protein)^-1 (16-3-fold purification). Best results were obtained when ammonium sulphate fractionation was done after ion-exchange chromatography rather than vice versa, since residual ammonium sulphate interfered with the ion-exchange chromatography when the fractionation was done before the ion-exchange step.

The 30-45% saturation ammonium sulphate precipitate was dissolved in 2 ml of TM buffer, dialysed and fractionated by molecular sieving through a Sephacryl S-200 column. L-Asparaginase eluted as a clear symmetrical peak when the column was eluted with TM buffer. This step resulted in good purification (Table 1) with an accumulated purification factor of 98. This purified preparation showed a major band of protein migrating in SDS-PAGE with a mobility corresponding to an 

$M_r$ and kinetics
The 

$M_r$ was determined more precisely by gel filtration. The $K_{sv}$ value of 0.22 obtained was slightly smaller than the observed value for BSA and corresponded to an 

$M_r$ of 80000 ± 1000, in good agreement with the value estimated by SDS-PAGE.

Normal saturation kinetics were obtained using L-asparagine as substrate in a Lineweaver-Burk double reciprocal plot. The apparent $K_m$ of L-asparaginase for L-asparagine was 2.5 mM. Neither D-asparagine nor L-glutamine nor any other related amino acid was a substrate for the enzyme. Neither of the two products of the reaction (aspartic acid and ammonium ion) exerted any inhibitory effect on enzyme activity up to a concentration of 100 mM.

Effect of DONV and other amino acids, and of cations on asparaginase activity
None of the 20 natural amino acids had any effect on L-asparaginase at concentrations up to 25 mM, even when the amino acid being tested was combined in a pair with aspartic acid, lysine, methionine, threonine, leucine, isoleucine, diaminopimelic acid or l-aminoethylcysteine. DONV, a structural analogue of L-asparagine (Handschumacher et al., 1968), inhibited L-asparaginase activity; the $K_i$ was 2.7 mM. Mg$^{2+}$, K$^+$ and other monovalent and divalent cations did not affect enzyme activity at concentrations up to 150 mM.

Isolation of pure L-asparaginase by preparative gel electrophoresis
The purified (98-fold) enzyme preparation showed two bands after staining with Coomassie brilliant blue (Fig. 1). Only the faster-moving of the two bands gave a positive L-asparaginase reaction, forming an ochre colour when incubated with hydroxylamine and stained with the iron chloride reagent (Fig. 1).

Both protein bands were electroeluted separately from the gel. In vitro, only the protein eluted from the faster-moving band gave a positive L-asparaginase reaction.

A sample of the eluted protein that had L-asparaginase activity was denatured with SDS. On SDS-PAGE this protein moved as a single band with an 

$M_r$ of 81000.

Discussion
L-Asparaginase is present in different forms in the microorganisms in which it has been studied. In the Achromobacteriaceae, a single protein has been reported
or with the constitutive asparaginase of \textit{E. coli}. The latter seems to be an overflow enzyme converting excess asparagine in the cytoplasm to aspartic acid in contrast to the periplasmic enzyme which probably acts as a scavenger of asparagine in the extracellular fluid. Mutants of \textit{E. coli} defective in cytoplasmic asparaginase have been isolated. This mutation is located in the \textit{asnA} locus and has been mapped at minute 39 of the \textit{E. coli} chromosome map (Del Casale et al., 1983; Bachmann, 1983).

We found similar levels of enzyme activity – about 20 mU (mg protein)$^{-1}$ in either complex or minimal media; the value did not change significantly with the age of the culture suggesting that the L-asparaginase of \textit{C. glutamicum} might be constitutive, as in \textit{V. succinogenes}. The pure enzyme had no apparent cofactor requirement, was relatively stable and was not affected by ions or amino acids. These results are of interest in the context of amino acid biosynthesis in \textit{C. glutamicum}, which overproduces lysine and threonine (Ishino et al., 1984). Both of these amino acids are derived from aspartic acid, which we have identified as the limiting precursor \textit{in vitro} of the pathway (unpublished results). L-Asparaginase is present in many of the peptones and proteins used in culture media for production of amino acids. Furthermore, L-asparagine may be formed in excess, especially in media supplemented with ammonium salts or urea which are used industrially both as nitrogen sources and as buffering agents. Under these conditions the L-asparaginase of \textit{C. glutamicum} probably acts as an overflow enzyme converting asparagine into aspartic acid.

The similarity of the \textit{C. glutamicum} enzyme to asparaginase I of \textit{E. coli} is further supported by the sensitivity of both enzymes to inhibition by DONV, a substrate analogue. Finally, the L-asparaginase of \textit{C. glutamicum} could have antitumour activity similar to those of other asparaginas. The availability of a new asparaginase for comparative structure–activity relationships in antitumour studies may be of great interest.

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\textbf{References}


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