Regulation of the cAMP Level in the Yeast *Saccharomyces cerevisiae*: the Glucose-induced cAMP Signal Is Not Mediated by a Transient Drop in the Intracellular pH

By JOHAN M. THEVELEIN,'* MONIQUE BEULLENS,' FANNY HONSHOWEN,' GREET HOEBEECK,' KATRIEN DETREMERIE,' BERNHARD GRIEWEL,2 JAN A. DEN HOLLANDER3 AND ARNOLD W. H. JANS2

1 Laboratorium voor Cellulaire Biochemie, Katholieke Universiteit te Leuven, Kardinaal Mercierlaan 92, B-3030 Leuven – Heverlee, Belgium
2 Max-Planck-Institut für Systemphysiologie, Rheinlanddamm 201, D-4600 Dortmund 1, FRG
3 Philips Medical Systems, PO Box 218, 5600 MD Eindhoven, The Netherlands

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Addition of glucose to derepressed cells of the yeast *Saccharomyces cerevisiae* is known to cause a rapid, transient increase in the cAMP level, which lasts for 1–2 min and induces a cAMP-dependent protein phosphorylation cascade. The glucose-induced cAMP signal cannot be explained solely on the basis of an increased ATP level. Transient membrane depolarization and transient intracellular acidification have been suggested as possible triggers for the cAMP peak. Addition of glucose to cells in which the plasma membrane had been depolarized still produced the increase in the cAMP level excluding membrane depolarization as the possible trigger.

Using in vivo $^3$P NMR-spectroscopy we followed phosphate metabolism and the time course of the drop in the intracellular pH after addition of glucose with a time resolution of 15 s. Under aerobic conditions the initial pH and ATP level were high. On addition of glucose, they both showed a rapid, transient drop, which lasted for about 30 s. Under anaerobic conditions, the initial pH and ATP level were low and on addition of glucose they both increased relatively slowly compared to aerobic conditions. Several conditions were found in which the pH drop which occurs under aerobic conditions could be blocked completely without effect on the cAMP signal or without completely preventing it: addition of NH$_4$Cl together with glucose at high extracellular pH and addition of a low concentration of glucose before a high concentration. Also, when glucose was added twice to the same cells no consistent relationship was observed between the pH drop and the cAMP peak. These results appear to exclude transient intracellular acidification as the trigger for the cAMP signal. Hence, we conclude that the effect of glucose cannot be explained on the basis of effects known to be caused by the membrane depolarizing compounds which cause increases in the cAMP level. A new, more specific kind of interaction appears to be involved.

INTRODUCTION

Addition of glucose to derepressed yeast cells causes a rapid transient increase in the cAMP level (Van der Plaat, 1974) which triggers a cAMP-dependent protein phosphorylation cascade similar to the well-known, hormone-induced phosphorylation cascades in mammalian cells (for reviews see Thevelein, 1984; Holzer, 1984). Based on the known effects of membrane depolarizing compounds two hypotheses have been proposed for the mechanism triggering the glucose-induced cAMP signal. The first was based on the observation that addition of glucose

**Abbreviation:** DNP, 2,4-dinitrophenol.
itself causes a transient depolarization of the plasma membrane (Kovac & Varecka, 1981; Mazon et al., 1982), and this was suggested to be the trigger for the transient cAMP increase (Mazon et al., 1982; Holzer, 1984). The second was based on the fact that the membrane depolarizing compounds that enhance the cAMP level in yeast also lower the intracellular pH. Because of the different pH optima of adenyl cyclase (pH 6) and phosphodiesterase (pH 8) a lowering of the intracellular pH was expected to raise the cAMP level (Londesborough, 1977). Since addition of glucose to (glucose-repressed) yeast cells causes a transient drop in the intracellular pH (den Hollander et al., 1981), it was suggested that a similar drop in derepressed yeast cells might trigger the glucose-induced cAMP signal (Busa & Nuccitelli, 1984). In recent publications (Caspani et al., 1985; Valle et al., 1986; Purwin et al., 1986) this hypothesis was supported on the basis of the similarity between the glucose effect and the effect of agents that lower the intracellular pH in yeast.

In the present paper we present evidence against both hypotheses. The effect of glucose is shown to be unrelated to the effect of membrane depolarizing compounds and we suggest that it involves a different mechanism.

METHODS

Yeast strain and culture conditions. The Y55 yeast strain and the culture conditions used were the same as described by Thevelein et al. (1987).

Incubation conditions. The incubation conditions were basically as described by Thevelein et al. (1987). Cells were suspended (300 mg wet wt ml⁻¹) in 25 mM-MES/NaOH buffer, pH 6, which was bubbled with pure O₂ ('aerobic conditions') or pure N₂ ('anaerobic conditions'). The incubation conditions in the cAMP determination experiments and in the NMR experiments were the same. It was technically not possible to take samples for cAMP determination during the NMR experiments. The experiments labelled 'shake culture' were done at low cell density as described by Thevelein et al. (1987). In all experiments, the cells were preincubated at 25°C for 10 min under the appropriate conditions. Incubation at high external pH was done in 25 mM-Tris/HCl buffer, pH 8 or 9.

CAMP determination. This was done as described by Thevelein et al. (1987).

In vivo ³¹P-NMR spectroscopy. In vivo NMR spectra were taken as a function of time as described by Thevelein et al. (1987).

Reproducibility of results. All experiments were repeated at least twice with consistent results. Representative results are shown.

RESULTS

To investigate whether or not the glucose-induced cAMP increase could be explained by the increased ATP level, we did experiments in which cells kept under anaerobic conditions were suddenly supplied with oxygen, or oxygenated cells were given glycerol or acetate (at pH 6). In none of these cases was a significant increase in the cAMP level observed (results not shown), indicating that under aerobic conditions the ATP level is not limiting cAMP synthesis. Evidence has been presented, however, that under anaerobic conditions the ATP level might be limiting (Thevelein et al., 1987). The glucose-induced cAMP increase was also significantly retarded under anaerobic conditions compared to aerobic conditions (Fig. 1) and this correlated with the slower increase in the ATP level (Fig. 3a, b).

Dinitrophenol (DNP) causes an increase in the cAMP level in yeast (Trevillyan & Pall, 1979; Mazon et al., 1982; Thevelein et al., 1987). It was shown to depolarize the plasma membrane in our yeast strain and under our experimental conditions (Thevelein et al., 1987). When DNP was added at pH 6-25, a moderate increase in the cAMP level was obtained (cf. Thevelein et al., 1987). Addition of glucose to the same suspension caused a second and much greater increase in the cAMP level (Fig. 2). The experiment shown in Fig. 2 was done in shake culture at low cell density. In similar experiments done under aerobic and anaerobic conditions at high cell density, addition of glucose to a culture treated with DNP caused a second rapid increase in the cAMP level.

Using ³¹P-NMR spectroscopy we investigated phosphate metabolism and the time-course of the intracellular pH change after addition of glucose to derepressed yeast cells under aerobic
Fig. 1. cAMP level after addition of 100 mM-glucose under aerobic and anaerobic conditions to a suspension of acetate-grown yeast cells in 25 mM-MES/NaOH buffer, pH 6. ●, Aerobic conditions; ▲, anaerobic conditions; △, shake culture.

Fig. 2. cAMP level after addition of 2 mM-DNP at pH 6.25 and subsequent addition of glucose (100 mM after 5 min, indicated by the arrow) to the same suspension.

Fig. 3. $^{31}$P-NMR spectra of acetate-grown yeast cells before (time zero) and after addition of glucose under aerobic (a) and anaerobic (b) conditions. Spectra were accumulated in blocks of 15 s; the spectra shown represent the mean value after 7.5 s, 22.5 s etc. The following peaks can be seen: P, free phosphate; ATP$, ATP_\beta$, ATP_, $\alpha$, $\beta$ and $\gamma$ phosphate groups of ATP; polyphosphate ($n \geq 4$); sugar phosphate [this broad peak is mainly composed of three components: glucose 6-phosphate, fructose 6-phosphate and fructose 1,6-bisphosphate (from left to right)]; NAD; the peak labelled 3PGA(?) probably corresponds to 3-phosphoglyceric acid (after addition of glucose this peak can be seen as a quickly disappearing shoulder at the right of the sugar phosphate peak).
The initial pH was high (Fig. 3a). Addition of glucose caused a transient drop (by about 70%) in the ATP level and (by about 75%) in the level of free phosphate. This coincided with a rapid buildup of sugar phosphates. The ATP level rapidly regained its initial value. Under anaerobic conditions, the initial ATP level was very low and increased rather slowly on addition of glucose. The buildup of sugar phosphates and the decrease of the phosphate level was clearly slower than under aerobic conditions (Fig. 3b).

From the position ('chemical shift') of the phosphate peak in the 31P-NMR spectrum, the intracellular pH can be determined (Gillies et al., 1982). The course of the intracellular pH after addition of glucose under aerobic and anaerobic conditions is shown in Fig. 4. Under aerobic conditions the initial pH was about 6.7-7.0; on addition of glucose it showed a relatively small but reproducible decrease which lasted for about 30 s. Under anaerobic conditions the initial pH was significantly lower (about 6.0-6.2); on addition of glucose it tended to decrease somewhat further before it increased to values similar to those present under aerobic conditions. This increase in the intracellular pH lagged somewhat behind the increase in the ATP level, under both aerobic and anaerobic conditions. Under anaerobic conditions, the intracellular pH before and just after glucose addition was much lower than the intracellular pH under aerobic conditions, even at the lowest value of the pH drop. The magnitude of the pH decrease observed under aerobic conditions was similar in different experiments. The initial pH present before the addition of glucose, however, was somewhat variable (e.g. see Fig. 5a). Therefore, and because of the small size of the pH drop, when the initial pH was rather high, the pH did not fall much below the initial pH observed in experiments where the initial pH was rather low (results not shown).

When 50 mM-NH₄Cl was added together with glucose at an external pH of 8 (Fig. 5a, aerobic conditions) or pH 9 (results not shown) the pH drop was completely abolished. Apparently, this is due to the rapid influx of NH₃ molecules which quickly become protonated once they are inside the cell because of the much lower intracellular pH. Lower concentrations of NH₄Cl (10 and 25 mM) did not prevent the pH drop completely. Fig. 5(a) also shows the variation of the initial intracellular pH between two duplicate experiments (before the addition of glucose, indicated by the arrow). Under conditions where the pH drop is completely abolished the cAMP response is delayed somewhat but certainly not completely inhibited (Fig. 5b). The delay might be due to unspecific inhibition of the cAMP synthesizing system by NH₃ or NH₂⁻. It has been observed that many positively charged compounds inhibit cAMP synthesis in vivo (Francois et al., 1984; Thevelein & Beullens, 1985). Anaerobically, a rapid increase in the intracellular pH took place when glucose was added together with 50 mM-NH₄Cl (Fig. 6a). Once again, the effect of NH₄Cl was limited to a short delay in the initiation of the cAMP peak (Fig. 6b). (When similar experiments were done in shake culture at low cell density, inhibition by NH₄Cl was more pronounced than under conditions identical to the NMR conditions. The reason for this is not known.)
The pH drop, but not the cAMP increase, could also be prevented by adding a low concentration of glucose just before a high concentration. This is shown in Fig. 7 for 5 mM-glucose followed 1 min later by 20 mM-glucose: 5 mM-glucose induced a small cAMP increase, 20 mM a much larger one (Fig. 7b); for the size of the pH drop in the two cases, the situation is just the opposite (Fig. 7a). Experiments with slightly differing conditions such as 8 mM-glucose followed by 25 mM, 10 mM followed by 25 mM and 5 mM followed by 10 mM followed by 100 mM, yielded results similar to those shown in Fig. 7.

Repeated addition of glucose to the same cells, washed and suspended each time in fresh buffer, causes new cAMP peaks which are smaller than the first one (Thevelein & Beullens, 1985). Fig. 8(a) shows the pH drop in cells which were given glucose twice. After the first glucose addition, the cells were washed and resuspended in fresh buffer for about 5 min. The initial pH before the second glucose addition was always higher, presumably because the cells were better energized. It can be seen from Fig. 8(a) that the second pH drop does not fall below the initial pH present before the first glucose addition. This is difficult to reconcile with the pH drop being the trigger for the glucose-induced cAMP peak since the peak was always observed after the second addition of glucose. This is shown in Fig. 8(b) for an experiment done under the same conditions as the NMR experiment. That the increased initial pH is not the cause of the lower second cAMP peak is corroborated by the fact that when, for each peak, DNP was added together with glucose (to lower the pH artificially to very low values), the second peak was still smaller than the first one (Fig. 8b). The decrease of the cAMP response therefore appears to be
Fig. 7. Course of the intracellular pH (a) and the cAMP level (b) after addition of 5 mM-glucose at time zero (first arrow) followed by 20 mM-glucose after 1 min (second arrow) under aerobic conditions.

Fig. 8. Reversibility of the glucose-induced pH drop (a) and the glucose-induced cAMP peak (b) under aerobic conditions. Glucose was added at time zero and the cells left in glucose for 60 min. The intracellular pH and the cAMP levels were measured during the first minutes. After 60 min, the cells were quickly chilled and washed with cold 25 mM-MES/NaOH buffer, pH 6, before the second glucose addition. In (b) a replicate culture was given 100 mM-glucose + 2 mM-DNP (○) instead of 100 mM-glucose (●).

due to inactivation of the cAMP synthesizing system (or activation of its breakdown system).

Under anaerobic conditions similar results were obtained for the pH drop (Fig. 9a) and for the reversibility of the cAMP response (Fig. 9b). In this case, however, straightforward interpretation of the data is not possible because of possible limitation of cAMP synthesis by the low ATP level just before and just after addition of glucose. In addition, the difference in the initial pH might be important for at least part of the decrease in the cAMP peak. It is also possible that glucose causes changes in the cAMP synthesizing system, such as a shift in the pH optimum.

DISCUSSION

Our results confirm that the glucose-induced cAMP increase cannot be explained solely on the basis of the glucose-induced increase in the ATP level. Oxygenation of derepressed yeast cells does not cause an increase in the cAMP level, although it produces ATP levels which are about the same as those observed after addition of glucose (Fig. 3a). Apparently, the cells possess enough endogenous substrates to maintain such a high ATP level for at least 20 min when supplied with pure oxygen (results not shown). It cannot be excluded, however, that the glucose-induced increase in the ATP level under anaerobic conditions is necessary for the cAMP
Fig. 9. Reversibility of the glucose-induced pH drop (a) and the glucose-induced cAMP peak (b) under anaerobic conditions. Glucose was added at time zero and the cells left in glucose for 60 min. The intracellular pH and the cAMP level were measured during the first minutes. After 60 min, the cells were quickly chilled and washed with cold 25 mM-MES/NaOH buffer, pH 6, before the second glucose addition. In (b) a replicate culture was given 100 mM-glucose + 2 mM-DNP (○) instead of 100 mM-glucose (●).

increase to occur. This might be the cause of the short delay in the increase under anaerobic conditions (Fig. 1). In the accompanying paper (Thevelein et al., 1987) we provided evidence that the low ATP level under anaerobic conditions in the absence of glucose might be limiting for cAMP synthesis.

Two hypotheses have been proposed in the literature for the mechanism by which glucose induces the cAMP increase in derepressed yeast cells: the first one was based on the transient depolarization of the plasma membrane which occurs on addition of glucose (Mazon et al., 1982; Holzer, 1984); the second on the transient decrease of the intracellular pH (Busa & Nuccitelli, 1984). We have presented evidence against membrane depolarization as a possible trigger for increases in the cAMP level (Thevelein et al., 1987). This is now confirmed for the glucose-induced increase in the cAMP level (Fig. 2). Membrane depolarization before addition of glucose did not prevent the rapid increase in the cAMP level. Since addition of DNP instead of glucose also produces a cAMP increase under these conditions (Thevelein et al., 1987), the ATP level in the depolarized cells cannot have been a limiting factor.

The evidence that the stimulating effect of proton ionophores on the cAMP level is due to intracellular acidification (Thevelein et al., 1987), appeared to support the hypothesis of Busa & Nuccitelli (1984) that a transient decrease in the intracellular pH could be the trigger for the cAMP signal. Addition of glucose to derepressed yeast cells causes a transient pH drop as had been found by den Hollander et al. (1981) for glucose-repressed S. cerevisiae and by Nicolay et al. (1982) for Zygosaccharomyces bailii. Recently, Caspani et al. (1985) and Valle et al. (1986) also provided evidence for the existence of a glucose-induced pH drop in derepressed S. cerevisiae cells, using classical pH determination methods. The pH drop is probably caused by the first and the third reactions of glycolysis, which are both proton-producing reactions. Apparently, the sudden massive initiation of glycolysis, as reflected by the rapid buildup of sugar phosphate,
causes a partial depletion of the ATP level and concomitant generation of sufficient protons to acidify the cells. Recovery of the pH is probably brought about by the plasma membrane H+-ATPase, which is itself also activated on addition of glucose to derepressed yeast cells (Serrano, 1983). Under anaerobic conditions the initial pH and the pH just after glucose addition were much lower than whatever pH under aerobic conditions. Under these conditions, however, cAMP synthesis is probably limited by the low ATP level.

Although the pH drop was reproducible with our strain, it was rather small. The magnitude of the pH drop apparently varies for different strains (see Valle et al., 1986). This might be due to differences in the vigour of the strains used, a more vigorous strain being able to recover faster from intracellular acidification than a less vigorous one. The activity of the plasma membrane ATPase and its degree of glucose-induced activation are probably crucial for this pH recovery. As a consequence of the small size of the pH drop and because of the variability of the initial pH of the cells before the addition of glucose, the lowest pH obtained in some experiments was not much below the initial pH present in other experiments. In addition, several conditions were found in which the pH drop was abolished without much effect on the cAMP peak, or in which there was no consistent relationship between the pH effect and the cAMP increase. This was true for addition of glucose together with NH$_4$Cl at high extracellular pH (Fig. 5a), addition of a small amount of glucose before a larger concentration (Fig. 7a) and repeated addition of glucose to cells washed free from glucose (Fig. 8a).

Our results indicate that the glucose-induced cAMP peak in derepressed yeast cells cannot be explained on the basis of known effects of membrane depolarizing compounds, i.e. neither membrane depolarization nor intracellular acidification. This was tried before by several authors (Mazon et al., 1982; Holzer, 1984; Busa & Nuccitelli, 1984; Caspani et al., 1985; Valle et al., 1986). Hence the effects of glucose and of proton ionophores do not share a common mechanism. Instead, we suggest that the effect of glucose is based on a more specific interaction which might be analogous to the mechanism involved in the generation of the hormone-induced cAMP signal in mammalian cells. Apparently, one of the components of the mechanism is glucose repressible: glucose-repressed yeast cells do not show a cAMP peak when given glucose (unpublished results). This component also appears to be inactivated rapidly on addition of glucose (Figs 8b and 9b; Thevelein & Beullens, 1985). Hence, the cAMP signal itself is probably controlled by glucose-induced inactivation (also called 'catabolite inactivation': Holzer, 1976), just like the enzymes for which it triggers inactivation (e.g. fructose bisphosphatase).

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REFERENCES


**Kovac, L. & Varecka, L. (1981).** Membrane poten-
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...tials in respiration-deficient yeasts monitored by a fluorescent dye. *Biochimica et biophysica acta* 637, 209–216.


