Most geothermal environments are rich in reduced sulfur compounds and provide suitable habitats for various groups of sulfur-oxidizing bacteria. However, in hydrothermal springs, growth of aerobic microorganisms is limited by the low solubility of oxygen at high temperatures. Other limitations may be imposed by competition with abiotic oxidations of sulfur compounds or intermediate metabolites. Facultatively or obligately aerobic sulfur oxidizers that grow at temperatures up to 90°C have been found to belong primarily to the archaeal domain (47, 54), particularly to the genera *Sulfolobus*, *Acidianus*, and *Desulfovorobius*.

According to Aragno (2), the thermophilic, aerobic, sulfur-oxidizing bacteria can be divided into three categories: hydrogen-oxidizing bacteria which are also able to oxidize reduced sulfur compounds (1, 4, 10) or require them (44), strictly thermophilic sulfur oxidizers belonging to the genus *Thermothrix* (16), and moderately thermophilic thiobacilli (15, 21, 31, 53). So far, *Thermothrix thiopara* (16) is the only validly described, thermophilic, sulfur-oxidizing (eu)bacterium. A phylogenetic analysis of this organism is included in this paper. *T. thiopara* is facultatively autotrophic and facultatively anaerobic, and its temperature range for growth is 60 to 80°C (12-14). Until recently, this organism was thought to be the only true thermophile that is able to reduce carbon dioxide to organic carbon via the Calvin cycle.

In this paper we describe an aerobic, thermophilic, obligately chemolithoautotrophic, sulfur-oxidizing bacterium belonging to the genus *Thermothrix*, *Thermothrix azorensis* sp. nov.

**MATERIALS AND METHODS**

Sample collection, isolation procedures, and culture conditions. White filamentous bacterial mat material was collected from a hot spring located at the caldera at Furnas on Sao Miguel Island in the Azores. The samples were stored at room temperature for more than 1 month before isolation procedures were started by serially diluting them in medium containing thiosulfate (4 mM) and hydrogen sulfide (0.5 mM) and incubating the preparations at 73°C. This resulted in isolation of strain TMT (T = type strain). The medium which we used contained (per liter) 0.5 g of (NH₄)₂SO₄, 0.1 g of MgSO₄·7H₂O, 0.05 g of CaCl₂, 1.2H₂O, 0.11 g of K₂HPO₄, 0.085 g of KH₂PO₄, 0.42 g of NaHCO₃, 5 ml of a vitamin mixture (3), and 1 ml of a trace element solution (8). The vitamin and trace element solutions were filter sterilized separately and, after cooling, were added to the autoclaved salt solution. Sodium thiosulfate, potassium tetrathionate, sodium sulfide, elemental sulfur, and sodium bicarbonate were sterilized separately and added, after cooling, to the medium at the final concentrations indicated below. When the following organic compounds were used as potential substrates, they were sterilized by filtration: glucose, arabinose, galactose, fructose, lactose, maltose, lactate, pyruvate, citrate, formate, fumarate, acetate, mannitol, glycine, a mixture of glutamate and serine, and yeast extract. These compounds were added to the medium (concentration range, 0.5 to 5 gliter) just prior to inoculation.

Enrichment cultures and, ultimately, pure cultures were grown in either 17-ml screw-top tubes (containing 5 ml of culture medium) or 125-ml flasks (containing 75 ml of culture medium) that were submerged in a heated and covered water bath. For certain growth experiments jacketed chemostat vessels were used. These vessels were connected to a thermostatic circulator (Haake, Berlin, Germany) to maintain the desired temperature inside the vessel. Sterile air was bubbled continuously through the stirred cultures. Inocula (10%, vol/vol) were grown on medium containing 4 mM thiosulfate. In some experiments the medium was supplemented with sterilized elemental sulfur ("precipitated powder"; final concentration, 1 gliter; Fisher Scientific Co., Pittsburgh, Pa.).

A pure culture of *T. thiopara* was kindly provided by Daniel K. Branaan, Abilen, Tex., and was grown on the medium described by Caldwell et al. (16).

**Analyses.** To determine numbers of cells, cells were fixed in 0.5% glutaraldehyde in sterile phosphate buffer, stained with 0.01% acridine orange, and counted by epifluorescence microscopy (22). Gram staining was performed with a Gram Stain Set kit (Difco). To determine sensitivity to lysozyme, concentrated suspensions of freshly grown cells (12 mg of cell protein per ml) in 0.1 M PiPES [piperezine-N,N'-bis(2-ethanesulfonic acid)] buffer (pH 7.0) were treated with lysozyme (final concentration, 1 mg/ml) for 30 min at 37°C. The amount of cell lysis was determined microscopically.

Protein content was determined by assaying a preparation with Coomassie blue (11) following hydrolysis of cells in 1 N NaOH at 100°C for 10 min. The concentrations of thiosulfate and tetrathionate were determined by the cyanoanalysis method (25), the concentration of sulfide was determined by the method of Cline (17), the concentration of sulfate was determined by using fuchsin formaldehyde (49), the concentration of sulfate was determined by using BaCl₂ and turbidimetry as described by Tabatabai (48), and the concentration of elemental sulfur was determined as described by Odintsova et al. (38).

**Microscopy.** Cells were examined by phase-contrast microscopy. Photographs
of living microorganisms were obtained by using wet mounts on agar-coated dry slides that were prepared as described by Pfennig and Wagener (39).

For transmission electron microscopy, cells were fixed with glutaraldehyde and osmium tetroxide, embedded in agar and in Spurr resin, and stained with uranyl acetate and lead citrate by the method of Waterbury and Stanier (52). Thin sections were examined with a Zeiss model 10 CA electron microscope.

**Preparation of cell extracts and cell suspensions.** The cells used for experiments performed with cell suspensions and enzyme assays were collected from late-log-phase cultures that were centrifuged at 10°C and 8,000 rpm for 15 min, washed twice in sterile substrate-free medium and then in 0.1 M PIPES buffer (pH 7.0), and recentrifuged. The resulting pellet was resuspended in the assay buffer. Extracts were prepared by passing cells suspended in the assay buffer three times through a French press at 7,000 lb/in² and centrifuging the preparation at 12,000 rpm for 20 min at 4°C. The resulting supernatant was used immediately in enzyme assays.

For short-term growth experiments, cell suspensions were prepared as follows.

After two washes the cells were resuspended in 0.25 M Tris-Cl buffer (pH 7.5), stirred for 1 h at 76°C, recentrifuged, and resuspended in 15 ml of the same buffer. Then the suspension was oxygenated by vigorous stirring with a stirring bar at 76°C for 40 min. At time zero, 0.15 g of sulfur was added.

**Lipid analysis.** Lipids were extracted, fractionated into neutral and polar lipid classes by using silicic acid, subjected to acid methanolysis to release the hydrophobic core lipids, and analyzed by thin-layer chromatography as described elsewhere (29, 30). To analyze and identify hydrophobic lipid components we used a Hewlett-Packard model 5890A gas chromatograph connected to a Hewlett-Packard model 3970 mass selective detector, as well as a Hewlett-Packard type HP-5 capillary column (10 m by 0.1 mm); electron impact spectra were obtained at 70 eV.

**Measurement of CO₂ fixation.** Time course experiments in which radiolabeled bicarbonate was used were performed by using 5 ml of preparation in 17-ml tubes with rubber stoppers. The medium which we used contained 0.1 μCi of NaHCO₃ per ml, 0.42 g of NaCl per liter, and 1 g of SO₄ per liter. The inoculum made up 10% of the volume. All of the components except SO₄ were mixed thoroughly before they were distributed into tubes. The sterile sulfur suspension was added separately to every tube. The experiments were carried out at 80°C. At certain times, two tubes were removed from the water bath and analyzed as described by Tuttle et al. (50).

Crude cell extracts were assayed for ribulose-1,5-biphosphate carboxylase/oxygenase activity by using the procedure of Beudeker et al. (7), as modified by Nelson and Jannasch (37). Fresh spinach extract was used as a control.

**Phylogenetic analyses.** DNA was isolated by the method of Marmur and Doty (33). The G+C content of the DNA was calculated from its melting point in 0.1X SSC by using salmon sperm DNA (41.2 mol% G+C) as the reference (1X SSC is 0.15 M NaCl plus 0.015 M sodium citrate). PCR amplification of the 16S rRNA gene was performed by using the following primers designed by using conserved bacterial sequences (36): forward primer EUBSFOR (5'-TGGAGAG'MTGATChlTGGCTCAG [rRNA numbering]) and reverse primer DREV (5'-ACGGNTACCITGTTACGACTT [rRNA numbering]; corresponding to positions 1512 to 1492). The sequences of PCR products were determined directly by using a Sequenase PCR product sequencing kit (United States Biochemical). The sequences obtained were manually aligned with closely related and representative sequences as determined with the utilities of the Ribosomal Database Project (31). Partial gene sequences were determined for strain TMT (825 nucleotides; levels of similarity with T. thiopara, 97 and 92%, respectively) and for T. thiopara (958 nucleotides; level of similarity with T. thiopara, 98%). Phylogenetic trees were generated by using the power PC version of the maximum-likelihood algorithm in the PHYLIP package of programs (20). Reference sequences were obtained from the Ribosomal Database Project (31).

**RESULTS**

**Morphology.** Exponentially grown cells of strain TMT under optimal conditions were gram-negative, long, thin rods that were 2 to 5 by 0.3 to 0.5 μm (Fig. 1). The cells were motile, often occurred in pairs, and were longer (up to 10 μm long) at temperatures above the optimum temperature (i.e., at temperatures above 78°C). Around 65°C, near the lower limit of the temperature range for growth, filaments up to 70 μm long were observed, especially at the beginning of the late-log phase. When oxidation of thiosulfate was incomplete for any reason and the pH was not below 7.0, cells often contained inclusions of elemental sulfur.

**General growth characteristics.** Strain TMT was isolated from serial dilutions in medium containing thiosulfate and hydrogen sulfide. Strain TMT grew aerobically and chemolithoautotrophically with hydrogen sulfide, thiosulfate, tetra-

thionate, or sulfur as the electron donor. It did not grow heterotrophically with any of the following organic compounds: glucose, arabinose, galactose, fructose, lactose, maltose, lactate, pyruvate, citrate, formate, fumarate, acetate, mannitol, glycine, a mixture of glutamate and serine, and yeast extract (concentration range tested, 0.5 to 5 g/liter). Growth on thiosulfate (0.5 to 1 g/liter) was not inhibited by glucose, acetate, lactate, mannitol, fumarate, arabinose, galactose, lactate, malto-

se, citrate, pyruvate, or yeast extract, and these organic compounds did not increase the cell yield during growth on thiosulfate (data not shown). Thus, strain TMT did not grow mixotrophically.
FIG. 2. Effect of temperature on the doubling time of strain TM' during growth on elemental sulfur.

When thiosulfate was replaced with methionine (0.5 g/liter) as the sulfur source, strain TM' did not grow in the presence or absence of organic compounds.

No growth occurred in the presence of hydrogen with or without sulfur or thiosulfate (the gas mixture contained 5% air). Growth did not occur anaerobically when thiosulfate, sulfur, or an organic compound was used as the electron donor and nitrate was used as the electron acceptor or when hydrogen was used as the electron donor and sulfur or thiosulfate was used as the electron acceptor. Strain TM' appeared to be an obligately aerobic autotroph.

Strain TM' grew well at 63 to 86°C, and the optimum temperature was 76 to 78°C. When this organism was grown on elemental sulfur at the optimum temperature, the doubling time of the culture was 1.5 h (Fig. 2). At 86°C the number of cells doubled only once, probably because of increased maintenance metabolism, but it was possible to subculture the organism at this temperature. When cells were grown on elemental sulfur, the typical lag phase was strongly affected by the temperature. The length of the lag phase decreased with increasing temperature.

Generally, strain TM' grew at pH 6.0 to 8.5, and the optimum pH was close to 7.1 (Fig. 3). At pH 8.5, filaments up to 30 μm long formed.

Strain TM' can be stored in liquid nitrogen with dimethyl sulfoxide, a method devised by Connaris et al. (18).

**Autotrophic growth of strain TM' on thiosulfate and sulfur.**

Growth of strain TM' on thiosulfate resulted in production of sulfate, and extracellular sulfur was an intermediate metabolite. No polythionates were detected. When the initial thiosulfate concentration was 18.25 mM, growth stopped when the concentration of the substrate was 8 mM, apparently because of a decrease in pH. The initial pH was 7.5, and the pH increased significantly during the first 1 h, reached a value of 9.2, and then decreased gradually to a value of 6.0 at the end of the growth period. When the initial concentration of thiosulfate in the culture was 7.5 mM, almost all of the substrate was used within 17.5 h; in this experiment the residue consisted of 0.1 mM thiosulfate, and the final pH was 5.6.

Independent of the initial pH of the medium, the pH increased to 8.7 to 9.0 in several experiments during the first 1 h of growth on thiosulfate or sulfur (Fig. 4). The pH then decreased gradually, leveling out at around 6.5. Growth on elemental sulfur appeared to be inhibited when 20 mM HEPES (N-2-hydroxyethylpiperazine-N'-2-ethanesulfonic acid) buffer was added to the medium. This indicated that the initial increase in pH may have favored activation of elemental sulfur as a substrate. Buffering did not affect growth on thiosulfate.

Growth on elemental sulfur under strictly aerobic conditions with constant mixing was preceded by a lag phase accompanied by production of sulfide and a rapid increase in pH (Fig. 4). Thiosulfate was the major end product, and sulfate was the minor end product. Under these conditions the thiosulfate
After an initial increase in the sulfide concentration for 40 min, when the pH was decreasing. The reason for the brief appearance of sulfide during the lag phase of the culture and the apparent absence of detectable sulfide thereafter might be that all of the sulfur was solubilized during the initial pH peak and then oxidized during the time when the pH was decreasing.

In order to distinguish between chemical and biological sulfide production from elemental sulfur in growing cultures, short-term experiments were conducted with suspensions of starved and washed (nongrowing) cells in Tris-HCl buffer (Fig. 6) at pH 7.5. Uninoculated controls were examined in parallel. After an initial increase in the sulfide concentration for 40 min, the sulfide concentration started to decrease while the thiosulfate concentration increased. In the uninoculated controls only an insignificant amount of sulfide was detected initially, while the level of thiosulfate increased during the first 20 min and decreased subsequently to 0.4 mM. These results support the notion that most of the initial sulfide production was indeed biological sulfide production.

**CO₂ fixation.** When strain TM³ was grown on sulfur at 80°C, the total uptake of radiolabeled sodium bicarbonate during a 9-h experiment was 17.45 μg of HCO₃⁻ fixed per 10⁶ cells. The average rate was 1.9 μg of HCO₃⁻, fixed per 10⁶ cells per h. This autotrophic activity was similar to the activities reported previously for other autotrophic bacteria. T. thiopara fixed carbon through the Calvin cycle (14), like most aerobic autotrophic eubacteria. However, activity of ribulose-1,5-biphosphate carboxylase, the key enzyme of the Calvin cycle, was not detected in our experiments. Likewise, ribulose-1,5-biphosphate carboxylase/oxygenase form I and II antibodies did not cross-react with a cell extract. The possibility that fixation of carbon occurs via the reductive tricarboxylic acid cycle is being investigated at this time.

**Chemotaxonomic characteristics.** Lipid analyses revealed that the hydrophobic core lipids of strain TM³ were composed solely of typical eubacterial n-C₁₄ to n-C₁₇ fatty acids, with n-C₁₆ and n-C₁₈ accounting for about 95% of the total. Archaeobacterial isoprenoid ether lipids, long-chain diols, nonisoprenoid diether, and the long-chain dicarboxylic fatty acids that are characteristic of several different thermophilic eubacterial groups were not present in strain TM³.

The DNA base composition of strain TM³ was 39.7 mol% G+C, which is the value obtained for T. thiopara. This value is similar to the DNA base compositions of bacteria belonging to the strictly chemolithoautotrophic hydrogen-oxidizing genera *Hydrogenobacter* and *Calderobacterium* (44).

The 16S rRNA sequence of strain TM³ placed this organism in the β subdivision of the Proteobacteria (Fig. 7), indicating that it is closely related to the genera *Alcaligenes* and *Pseudomonas*.

**DISCUSSION**

We found that strain TM³ is a thermophilic, obligately autotrophic, aerobic bacterium that grows on reduced sulfur compounds. This organism is gram negative and sensitive to lysozyme and has a lipid composition that is common among eubacteria. Its morphological features, including the ability to form filaments, are similar to those of *T. thiopara* (12, 14). However, strain TM³ differs from the only previously described *Thermothrix* species (*T. thiopara*) by being strictly chemolithoautotrophic; it does not grow heterotrophically under aerobic or anaerobic conditions. In addition, strain TM³ has higher optimum and maximum growth temperatures than *T. thiopara*, and the G+C contents of the DNAs of the two organisms are the same (39.7 mol%). Both organisms appear to belong to the same genus of thermophilic sulfur-oxidizing bacteria but different species.

Geothermal springs are often colonized by bacteria that produce streamers, which are visible as a network of filaments and single cells. Previous attempts to isolate these filaments usually resulted in cultures of rod-shaped bacteria that produced filaments only under certain conditions. The only truly filamentous thermophilic bacterium that has been described so far belongs to genus *Thermus* (23). *T. thiopara* produces filaments only when oxygen is the growth-limiting factor (14), and...
Acetothermus paucivoruns, an anaerobic thermophilic bacterium, forms filaments only in the absence of vitamin B₁₂ (19). Acetothermus cellulolyticus formed filaments when it was first isolated but lost this morphological characteristic after repeated transfers (35). Members of the genus Hydrogenobacter have been reported to form filaments sometimes when they are incubated at temperatures above the optimal growth temperature (2).

Strain TM₇ produced filaments under unfavorable conditions (e.g., when it was grown at a temperature near the maximum temperature and at extreme pH values). As in the case of Hydrogenobacter strains, filament formation in thermophilic bacteria may be a reaction to stress. Since the conditions in natural environments are variable and most of the time far from favorable, filamentous forms or streamers are observed most often in situ. In a recent phylogenetic study (40) of a well-known pink community of filamentous organisms in the 88°C outflow of Octopus Spring in Yellowstone National Park, it was shown that the filaments are most closely related to the hydrogen-oxidizing bacterium *Aquifex pyrophilus* and its close relative *Hydrogenobacter thermophilus*. When these organisms were grown in culture, however, both of them occurred as unicellular bacteria.

An unusual observation was the initial reduction of elemental sulfur to sulfide in aerobic cultures of strain TM₇. This reduction occurred during the lag phase (i.e., when there was no increase in cell biomass). Reduction of elemental sulfur is common in anaerobic cultures of hyperthermophilic archaea and bacteria in which hydrogen, the electron donor, occurs as a fermentation product. *Desulfurolobus ambivalens* and one of the *Sulfolobus* species are able to grow either anaerobically by reducing sulfur or aerobically by oxidizing sulfur depending on the oxygen supply (42, 55, 56). The sulfur oxygenase reductase of *D. ambivalens* simultaneously produces sulfite, thiosulfate, and hydrogen sulfide from sulfur in the presence of oxygen (26, 27). This enzyme was not detected in anaerobically grown cells. It is possible that a similar enzyme is responsible for sulfur utilization in cultures of strain TM₇.

Hydrogen sulfide was not detected in cultures of strain TM₇ after the lag phase (Fig. 4). It is possible that in this organism sulfide serves as a sulfur nucleophile, as defined by Blumentals et al. (9), and that it solubilizes elemental sulfur, as has been shown for the anaerobic hyperthermophilic archaeon *Pyrococcus furiosus* (9).

In order to utilize sulfur, both mesophilic and thermophilic bacteria must either have direct contact with the solid sulfur substrate or mediate conversion of the sulfur ring to a soluble compound. The most likely elemental sulfur solubilization reaction is formation of polysulfides (9, 41). However, in aqueous solutions polysulfides are not stable at pH values below 8.7, and it has been shown that no sulfur solubilization occurs at pH 7.0 (9). Thus, in cultures of strain TM₇ initial sulfide production from elemental sulfur should lead to the increase in pH necessary for sulfur solubilization.

Sulfide ions (S²⁻ and HS⁻) are probably the most important sulfur nucleophiles present in bacterial systems. The initial ring-opening reaction, 

\[ \text{S}_n + \text{HS}^- \rightarrow \text{S}_{n-1}\text{S}^2^- + \text{H}^+ \]

is followed by rapid chain degradation and establishment of an equilibrium among polysulfide ions of different sizes (S₀, S₁, and so on), as well as sulfide and hydroxyl ions, according to the following reaction (9): 

\[ \text{S}_{n-1}\text{S}^2^- + \text{HS}^- + \text{OH}^- \rightarrow \text{S}_n\text{S}^2^- + \text{S}^- + \text{S}_2^- + \text{H}_2\text{O} \]

At high temperatures and near neutral pH, polysulfides apparently are chemically produced as intermediates from sulfur reduction and are subsequently utilized biologically (41). Polysulfides have also been identified as soluble intermediates during sulfur respiration by *Wolinella succinogenes* (28). In this connection it should be remembered that the energy threshold for sulfur reduction appears to be lower at temperatures above 80°C and that under anoxic conditions sulfide is produced abiotically (6). In aerobic sulfur-oxidizing bacteria, however, initial sulfide production has not been reported previously, and this phenomenon certainly should be studied further. The sulfide produced in cultures of strain TM₇ during the lag phase could serve the following two purposes: it could increase the pH for polysulfide stability, and it could accelerate sulfur solubilization, making sulfur available for oxidation.

Until recently, it was thought that all gram-negative aerobic autotrophic bacteria use the enzymatic Calvin cycle for CO₂ fixation. However, the recent isolation and studies of thermophilic hydrogen-oxidizing bacteria belonging to the genera *Hydrogenobacter* and *Aquifex* revealed that the reductive tricarboxylic acid cycle also occurs in aerobic autotrophic eubacteria (5, 24, 43, 44). So far, this cycle appears to be restricted to these two genera, which form the deepest branch on the bacterial phylogenetic tree (40, 45). It is interesting, therefore, that strain TM₇, which is not related to the hydrogen-oxidizing thermophiles, also did not exhibit ribulose-1,5-biphosphate carboxylase/oxygenase activity. The closest relative of strain TM₇, *T. thiopara*, was originally reported to have ribulose-1,5-biphosphate carboxylase/oxygenase activity (14). Later, however, this could not be confirmed, and a C₄ pathway was suggested for CO₂ fixation (34). Besides a phylogenetic basis for the absence of the Calvin cycle in autotrophic thermophilic bacteria, a number of other factors, such as growth temperature, might be responsible.

The results of our phylogenetic analysis of partial rRNA sequences suggested that strain TM₇ and *T. thiopara* are closely related species that belong to the β subdivision of the *Proteobacteria* (formerly the purple bacteria and their relatives) (46). The genus *Thermothrix* appears to contain the only thermophilic bacteria in the β subdivision. Figure 7 shows the relationship of members of this genus to representatives of the various other subdivisions of the *Proteobacteria*. Although the DNA base composition of strain TM₇ is similar to the DNA base compositions of the thermophilic hydrogen-oxidizing bacteria, strain TM₇ can be separated from these bacteria on the basis of physiological properties and phylogenetic position.

**Description of *Thermothrix azorensis* sp. nov.**

*Thermothrix azorensis* Odintsova, Janasch, Mamone, and Langworthy (a.z.o.ren.sis. L. fem. adj. azorensis, from the Azores).

Cells are long thin rods that are 2 to 5 by 0.3 to 0.5 μm and often occur in pairs. Filaments up to 70 μm long are produced under unfavorable growth conditions. Cells are gram negative and nonsporulating and may contain sulfur inclusions. The organism is strictly aerobic and obligately chemolithoautotrophic and uses reduced sulfur compounds (hydrogen sulfide, elemental sulfur, tetrahydrothionate, and thiosulfate) as electron donors and carbon dioxide as a carbon source. During the lag phase oxidation of elemental sulfur is preceded by production of sulfide.

The temperature range for optimum growth is 76 to 78°C, and the temperature extremes are 60 and 87°C. Growth occurs at pH 6.0 to 8.5, and the optimum pH is 7.0 to 7.5.

The lipid fatty acids include n-C₁₀ to n-C₁₈ fatty acids, with n-C₁₀ and n-C₁₆ fatty acids accounting for 95% of the total fatty acids.

The G+C content of the DNA is 39.7 mol%. As determined by a 16S rRNA sequence analysis, our new isolate is affiliated with the β subdivision of the *Proteobacteria*.

*T. azorensis* was isolated from a hot spring located at the caldera at Furnas, Sao Miguel, the Azores, Portugal.
The type strain is strain TM, which has been deposited in the American Type Culture Collection as strain ATCC 51754.

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